

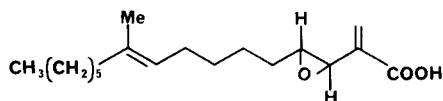
A GENERAL METHOD FOR THE SYNTHESIS OF 3,4-EPOXY-2-METHYLENEALKANOIC
ACID DERIVATIVES USING DIANION OF *N*-PHENYL-2-(PHENYLSULFONYLMETHYL)-
PROPENAMIDE

Kazuhiko Tanaka,* Hidemi Yoda, and Aritsune Kaji
Department of Chemistry, Faculty of Science, Kyoto University,
Sakyo, Kyoto 606, Japan

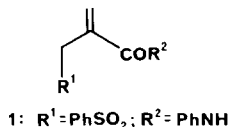
Summary: A general procedure has been developed for the synthesis of
3,4-epoxy-2-methylenealkanoic acid derivatives using dianion of
N-phenyl-2-(phenylsulfonylmethyl)propenamamide and aldehydes.

The α -methylene carbonyl system is a common structural feature of naturally occurring substances possessing cytotoxic, fungitoxic, and growth-inhibitory activity. Although various methods have been reported for the preparation of α -methylene carbonyl compounds, little attention has been given to the synthesis of unusual structural units such as conocandin,¹ isolated from a strain of *Hormococcus conorum*, which has an additional epoxy function at the β, γ -position of the carbonyl group as shown in the structure.² In this communication we describe a general method for the preparation of such structural system containing 3,4-epoxy-2-methylene carbonyl functionality.

Our synthetic strategy is based on the utility of dianion 2 because it shows high α -regioselectivity in the reaction with aldehydes (Scheme I). Thus, treatment of the dianion 2 with 1.1 equiv of nonanal in THF containing 2.1 equiv of HMPA at -78°C for 2 h gave, after silica gel-column chromatography, 4-hydroxy-2-methylene-*N*-phenyl-3-(phenylsulfonyl)dodecanamide (3a)³ in 78% yield. The hydroxy group of 3a was protected as a trimethylsilyl ether, and the resulting ether 4a⁴ was treated with 2 equiv of sodium benzeneselenolate in ethanol at 0°C for 2 h, giving exclusively (*E*)-*N*-phenyl-2-phenylselenomethyl-4-trimethylsilyloxy-2-dodecenamide (5a) in 96% yield after extractive isolation and chromatography.⁵

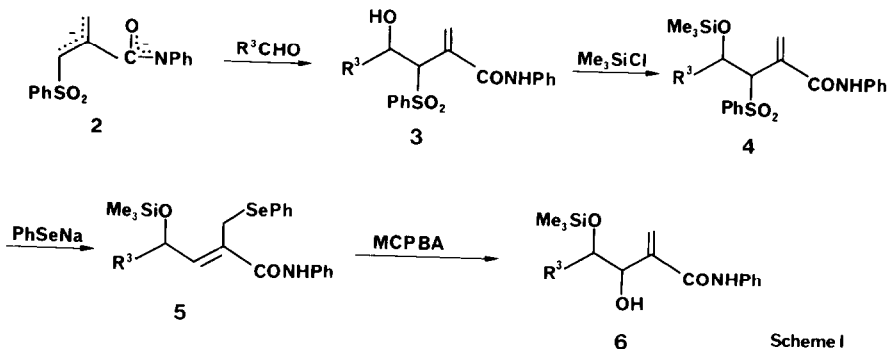


Conocandin



1: R¹=PhSO₂; R²=PhNH

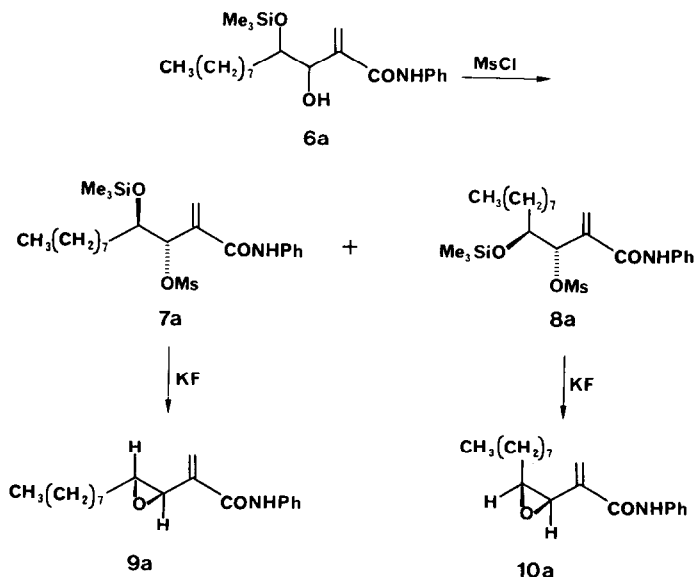
Oxidation of the amide 5a with 1 equiv of MCPBA in dichloromethane at -78°C for 1 h provided 1,2-diol derivative 6a in 88% yield.⁶ The results of other examples are summarized in Table 1.



Reaction of 3-hydroxy-2-methylene-*N*-phenyl-4-(trimethylsilyloxy)dodecanamide (6a) with 1 equiv of methanesulfonyl chloride gave a mixture of stereoisomers 7a (41%) and 8a (27%) which were easily separated by open column chromatography on silica gel using hexane-ethyl acetate (4:1) as eluent.⁷

Table 1. Preparation of Amides 3-6 (See Scheme I)

aldehyde	3 (yield)	4 (yield)	5 (yield; <i>E/Z</i> ratio)	6 (yield)
$n\text{-C}_8\text{H}_{17}\text{CHO}$	3a (78%)	4a (83%)	5a (96%; 100/0)	6a (88%)
$(\text{CH}_3)_2\text{CHCH}_2\text{CHO}$	3b (56%)	4b (82%)	5b (91%; 94/6)	6b (98%)
$c\text{-C}_6\text{H}_{11}\text{CHO}$	3c (77%)	4c (76%)	5c (87%; 100/0)	6c (75%)

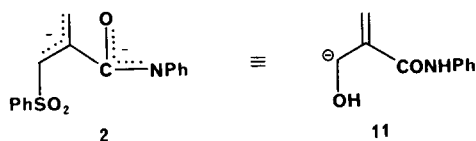


Conversion of trimethylsilyl ethers 7a and 8a into epoxides 9a and 10a was achieved in one step. Thus, reaction of erythro isomer 7a, the more mobile amide, with 2 equiv of potassium fluoride in DMSO afforded exclusively trans-3,4-epoxy-2-methylene-*N*-phenyldodecanamide (9a) in 66% yield.⁸ The less mobile amide, threo isomer 8a, upon treatment as described above gave cis-3,4-epoxy-2-methylene-*N*-phenyldodecanamide (10a) in 66% yield.⁸ The structural assignments of 9a and 10a were based on the analysis of ¹H NMR spectra. The small vicinal coupling constant ($J = 2.5$ Hz) of 9a confirmed the trans relation of C(3)-H and C(4)-H, while the larger coupling constant ($J = 4.5$ Hz) of 10a was consistent with the cis stereochemistry.^{1,2,9}

Similarly, reaction of 3-hydroxy amide 6b with methanesulfonyl chloride afforded a separable mixture of 7b and 8b in a ratio of about 1:1 which, upon treatment with fluoride ion, furnished the corresponding 3,4-epoxy-2-methylene amides 9b (54%) and 10b (55%), respectively.



In these synthetic sequences, the dianion 2 is synthetically equivalent to anion 11.



Since secondary amides can be easily converted to carboxylic acids or esters under very mild conditions,¹⁰ the present method provides a wide variety of 3,4-epoxy-2-methylenealkanoic acid derivatives from the readily available starting materials.

References and Notes

1. J. M. Müller, H. Fuhrer, J. Gruner, and W. Voser, *Helv. Chim. Acta*, **59**, 2506 (1976).
2. a) L. Banfi, D. Potenza, and G. S. Ricca, *Org. Mag. Res.*, **22**, 224 (1984).
 b) L. Banfi, L. Colombo, C. Gennari, and C. Scolastico, *J. Chem. Soc., Chem. Commun.*, 1983, 1112.
 c) L. Banfi, A. Bernardi, L. Colombo, C. Gennari, and C. Scolastico, *J. Org. Chem.*, **49**, 3784 (1984).
- d) J.-P. Corbet and C. Benezra, *Tetrahedron Lett.*, 1979, 4003.

3. Compound 3a: $^1\text{H NMR}$ (CDCl_3) δ 8.19 (m, 1H), 7.71-7.89 (m, 2H), 6.89-7.60 (m, 8H), 6.26, 6.13 (s, 1H), 6.07, 5.72 (s, 1H), 4.20-4.72 (m, 2H), 3.40-3.95 (broad, 1H), 0.96-1.72 (m, 14H), 0.60-0.96 (m, 3H); IR (neat) 3320, 1650, 1600, 1310, 1155, 1090 cm^{-1} .
Anal. Calcd for $\text{C}_{25}\text{H}_{33}\text{NO}_4\text{S}$: C, 67.69; H, 7.50; N, 3.16. Found: C, 67.48; H, 7.27; N, 3.42.
4. Compound 4a: $^1\text{H NMR}$ (CDCl_3) δ 8.21 (broad, 1H), 6.82-7.90 (m, 10H), 6.12 (d, $J = 5$ Hz, 1H), 6.02 (d, $J = 5$ Hz, 1H), 4.28-4.94 (m, 2H), 0.62-1.94 (m, 17H), 0.15 (s, 9H); IR (neat) 1675, 1605, 1315, 1160, 1100 cm^{-1} . Anal. Calcd for $\text{C}_{28}\text{H}_{41}\text{NO}_4\text{SSi}$: C, 65.20; H, 8.01; N, 2.72. Found: C, 65.15; H, 7.86; N, 2.61.
5. Compound 5a: mp 67-68°C. Anal. Calcd for $\text{C}_{28}\text{H}_{41}\text{NO}_2\text{SiSe}$: C, 63.37; H, 7.79; N, 2.64. Found: C, 63.40; H, 7.64; N, 2.71.
6. Compound 6a: $^1\text{H NMR}$ (CDCl_3) δ 8.81, 8.59 (broad, 1H), 6.92-7.60 (m, 5H), 5.99 (d, $J = 7$ Hz, 1H), 5.54 (s, 1H), 4.36 (broad, 1H), 3.26-3.96 (m, 2H), 0.64-1.64 (m, 17H), 0.04 (s, 9H); IR (neat) 3280, 1660, 1600, 1085 cm^{-1} .
Anal. Calcd for $\text{C}_{22}\text{H}_{37}\text{NO}_3\text{Si}$: C, 67.47; H, 9.52; N, 3.58. Found: C, 67.17; H, 9.41; N, 3.56.
7. Erythro-isomer 7a: $^1\text{H NMR}$ (CDCl_3) δ 8.03 (broad, 1H), 6.89-7.45 (m, 5H), 5.92 (s, 1H), 5.63 (s, 1H), 5.22 (d, $J = 4$ Hz, 1H), 3.89-4.11 (broad, 1H), 2.83 (s, 3H), 0.85-1.75 (m, 14H), 0.51-0.85 (m, 3H), 0.01 (s, 9H); IR (neat) 3310, 1670, 1600, 1360, 1180 cm^{-1} .
Threo-isomer 8a: $^1\text{H NMR}$ (CDCl_3) δ 7.93 (broad, 1H), 6.88-7.52 (m, 5H), 5.94 (s, 1H), 5.64 (s, 1H), 5.19 (d, $J = 5$ Hz, 1H), 3.84-4.08 (m, 1H), 2.89 (s, 3H), 0.96-1.65 (m, 14H), 0.63-1.65 (m, 3H), 0.01 (s, 9H); IR (neat) 3340, 1670, 1600, 1360, 1185 cm^{-1} .
8. Trans epoxide 9a: $^1\text{H NMR}$ (CDCl_3) δ 8.68 (broad, 1H), 6.99-7.67 (m, 5H), 6.23 (s, 1H), 5.74 (s, 1H), 3.57 (d, $J = 2.5$ Hz, 1H), 3.10 (dt, $J = 2.5, 5$ Hz, 1H), 1.06-2.00 (m, 14H), 0.72-1.06 (m, 3H); IR (neat) 3270, 1660, 1630, 1600, 905, 760 cm^{-1} . Anal. Calcd for $\text{C}_{19}\text{H}_{27}\text{NO}_2$: C, 75.71, H, 9.03; N, 4.65. Found: C, 75.56; H, 9.05; N, 4.54.
Cis epoxide 10a: $^1\text{H NMR}$ (CDCl_3) δ 8.32 (broad, 1H), 6.98-7.64 (m, 5H), 6.26 (s, 1H), 5.64 (s, 1H), 3.92 (d, $J = 4.5$ Hz, 1H), 3.08-3.36 (m, 1H), 0.98-1.78 (m, 14H), 0.64-0.98 (m, 3H); IR (neat) 3280, 1660, 1640, 1600, 770 cm^{-1} .
9. M. Shimagaki, T. Maeda, Y. Matsuzaki, I. Hori, T. Nakata, and T. Oishi, Tetrahedron Lett., 25, 4775 (1984).
10. D. L. Flynn, R. E. Zelle, and P. A. Grieco, J. Org. Chem., 48, 2424 (1983).

(Received in Japan 18 July 1985)